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THE ATOMIC MASS 11. Use your knowledge of atomic calculations to complete the chart. Element Atomic Number Atomic Mass Protons Neutrons Electrons Li 3 4 3 P 15 16 15 Cl 17 18 17 Ni 59 28 28 K 19 19 20 Ag 108 47 47 H 1 0 1 Si 14 28 14 W 74 174 74 Ne 10 20 10 NOTE: The number protons and electrons is equal to the atomic number.

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in a neutral atom of that element. The atomic number gives the "identity" of an element as well as its location on the periodic table. No two different elements will have the atomic number. The Of an element is the average mass of an element's naturally occurring atom, or isotopes, taking into account the Of each isotope.

090412 Atomic Structure Worksheet 1

tells you the number of _____ in a neutral atom of that element. The atomic number gives the "identity" of an element as well as its location on the periodic table. No two different elements will have the _____ atomic number. 4.

Basic Atomic Structure Worksheet

This type of worksheet discusses about basic atomic structure. The clean worksheet uses plain text with the following questions to be answer in order to study atomic structure. Basically, this worksheet is composed of questions and table to be filled the missing items.

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To the left of the symbol we find the atomic mass (A) at the upper corner, and the atomic number (Z) at the lower corner. Examples: ${}^1_1\text{H}$ ${}^{12}_6\text{C}$ ${}^{23}_{11}\text{Na}$ A Symbol Z Electron trade constitutes the currency of chemical reactions. The number of electrons in a neutral atom (that is, the atomic number) gives the element its unique identity.

1. ATOMIC STRUCTURE FUNDAMENTALS

ATOMS: ATOMIC STRUCTURE QUESTIONS AND ANSWERS . QUESTION ONE: MODELS OF THE ATOM (2011;1) At different times scientists have proposed various descriptions or models of the atom to match experimental evidence available. (a) The model that Thomson proposed was called the plum-pudding model. Describe this model.

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forces created by charged particles, especially electrons, and the motion and interactions of those particles. The electron is a fundamental component of matter and is considered to have the smallest possible unit of negative charge ...

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